

$$\boxed{B3-3} \quad (a) \quad Q_A/Q_B = \frac{y}{1-y} = \frac{0.6}{0.4} = \underline{1.5}$$

$$(b) \quad 2000 \gamma - 100 \times 0.6 \text{ Pa} \\ = \underline{2 \times 10^6 \gamma - 6 \times 10^4 \text{ Pa}}$$

$$(c) \quad Q_A = K_A S \frac{P_{1A} - P_{2A}}{L} = K_A \frac{S}{L} \times (2.0 \times 10^6 \gamma - 6 \times 10^4)$$

$$Q_B = K_B S \frac{P_{1B} - P_{2B}}{L} = K_B \frac{S}{L} \times (2.0 \times 10^6 (1-\gamma) - 4 \times 10^4)$$

$$\frac{K_A}{K_B} = \frac{10^{-12}}{10^{-13}} = 10.$$

$$\frac{Q_A}{Q_B} = \frac{2.0 \times 10^6 \gamma - 6 \times 10^4}{2.0 \times 10^6 (1-\gamma) - 4 \times 10^4} \times 10 = 1.5.$$

$$\text{Solve for } \gamma \quad \underline{\gamma = 0.154}$$

$$(d) \quad \frac{Q_A}{S} = 10^{-12} \times \frac{2 \times 10^6 \times 0.154 - 6 \times 10^4}{10^{-6}} = \underline{0.248} \text{ mol/m}^2 \cdot \text{s}$$

$$\frac{Q_B}{S} = 10^{-13} \times \frac{2 \times 10^6 \times (1-0.154) - 4 \times 10^4}{10^{-6}} = \underline{0.165} \text{ mol/m}^2 \cdot \text{s}$$

$$\frac{Q}{S} = \frac{Q_A}{S} + \frac{Q_B}{S} = \underline{0.413} \text{ mol/m}^2 \cdot \text{s}$$

$$(e) \quad Q_{\text{tot}} = 1 \text{ mol/s} \text{ ("gas") } \quad Q_{FA} = 0.3 \text{ mol/s}, \quad Q_{FB} = 0.7 \text{ mol/s}$$

$$\text{So } Q_{0A} = 0.3 - 0.248 \text{ S}, \quad Q_{0B} = 0.7 - 0.165 \text{ S}$$

(c) f1)

$$\frac{Q_{0A}}{Q_{0B}} = \frac{0.3 - 0.248 \text{ S}}{0.7 - 0.165 \text{ S}} = \frac{0.154}{1 - 0.154}$$

$$\text{Solve for } \gamma \quad \underline{S = 0.79} \text{ m}^2$$